## Advanced Laser Processes for Photovoltaic Energy Production

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### Nanomaterials & Organic Electronics Group Center of Advanced Materials & Photonics



### **Nanomaterials & Organic Electronics Group**

#### **Center of Advanced Materials & Photonics**







Characterization







13



Fabricate Test





Formulate Process

### Organic & Hybrid Photovoltaics



#### Flexible electronics





### Outline



#### Introduction to Organic Photovoltaics

- Status of OPVs
- ➢ Why OPVs / Applications of OPVs
- Working principle
- Device Architectures
- > Approaches to improve OPV performance (efficiency and stability)

### Laser Processes in OPVs

- ≻Why Lasers?
- ➢Photothermal Processes
- Photochemical Processes

#### Photothermal Processes exploited in nano group

- Reduction of Graphene Oxide Films: Towards free ITO OPVs
- Laser ablation method for NPs production
- Graphene-based OPVs

#### Photochemical Processes exploited in nano group

- Laser Induced Doping
- Laser Production of metal NPs
- Laser Induced Decoration

### ✤ A brief review of our group: Research & Educational Activities



## **PV Technologies**



#### **First Generation**

#### crystalline silicon (mono & multi)

This will be the PV-backbone technology and leader of the BIPV sector.

module efficiency:



#### Second Generation

#### thin film: a-Si, CdTe, CIGS

Viable competitor in BIPV and roll-to-roll process for flexible substrates.

module efficiency:



### **Disruptive/ New Generation**



## **Cell efficiency**







< Incident-photon-to-current conversion efficiency (IPCE) >



- I<sub>SC</sub> : Short-circuit current  $\rightarrow$  Current value when V = 0 V<sub>OC</sub>: Open-circuit voltage → Voltage value when I = 0**P** : Power output of the cell P = IV $FF = \frac{I_{mp}V_{mp}}{I_{sc}V_{oc}}$ **F.F : Fill factor**  $\eta = \frac{I_{sc} V_{oc} FF}{P_s} \times 100$ Under AM 1.5G simulated solar illumination
  - no. of photons incident  $[1240 \, eV \, nm][photocurrent \, density \, (\mu A \, cm^{-2})]$ 
    - [wavelength (nm)][irradiance (mW cm $^{-2}$ )]







### **Best Research-Cell Efficiencies**

### 





## Why OPVs ?



- Processed easily over large area
  - -spin-coating
  - -doctor blade techniques (wet-processing)
  - -evaporation through a mask (dry processing)-printing
- ✓ Low cost of fabrication
  - ✓ Poly and TF-Si costs around 500-1000 \$/m<sup>2</sup>
  - ✓ Plastic-PV expected below 50 \$/m<sup>2</sup>
- Light weight technology
- Mechanical flexibility and transparency
- Band gap of organic materials can be easily tuned chemically by incorporating different functional groups







## What is special about OPVs applications?









## What is special about OPVs applications?



→ Flexible









## Inorganic vs Organic



### **Theory of Inorganic Semiconductors**

- p-n junction :
- Light creates electron-hole pairs throughout semiconductor
- p and n layers provide selective contacts
- Free charge carriers
- Mobility of charges is relatively high



## **Inorganic vs Organic** Theory of Organic Semiconductors



### **Conjugated Polymers:**

- Light creates bound e-h pairs, excitons (0.4eV)
- Mobility of charges is relatively low

In a single-layer polymer photovoltaic device, excitons may be dissociated at:

- strong internal electric fields found at polymer/metal interfaces.
- <u>dissociation centres</u>, such as oxygen acting as e<sup>-</sup> traps.







## **Road to Bulk Heterojunction concept in OPVs**







### **Blended Devices - Bulk Heterojunction Concept**



- Mix hole and electron acceptor materials with offset energy levels together  $\rightarrow$  bicontinuous nanoscale mixture of D-A materials
- Both D and A materials must be very soluble in the same solvent
- Length scale of blend ~ exciton diffusion length (10-20nm)
- Exciton dissociation at interfaces; Electrons transferred to e<sup>-</sup> acceptor, while h<sup>+</sup> to the polymer
- Percolation limit for both materials has to be reached.

The blend morphology has to enable charge-carrier transport in the two different phases in order to minimize acceptor recombination



## **OPV device operation principle**







## **Device structures**







## How to improve OPV performance - Research Activities



**Photon Management** 



#### **Plasmonics**

Metallic nanostructures enable light manipulation at the nanoscale. Photonic and plasmonic enhancement of photocurrent **Novel materials** 



Solution processable graphene and 2D based materials

As electrode, interfacial layer, electron acceptors, and additive materials, compatible with r2r

### **Device Engineering**



Ternary organic-inorganic hybrids etc.

Ternary devices to enhance light harvesting; Optimize electron/hole transport

Enhancing OPV Performance and Stability through Integrated Material, Interfacial, and Device Engineering







## **Utilize the Laser Tool**

Lasers represents an interesting fabrication tool in the field of OPVs. Lasers based photochemical & photothermal processes are fast, precise, fully automated, localized, facile and fully controllable







### **Photon Management**

Metallic nanostructures enable light manipulation at the nanoscale. Photonic and plasmonic enhancement of photocurrent





## **Photon Management**

### Key problems towards high efficient OPVs

- $\checkmark$  Exciton binding energy ~200-500meV
- Exciton diffusion length  $l \sim 10 \text{ nm}$  $\checkmark$
- $\checkmark$  1 << layer thickness
- Tradeoff between light absorption length  $\checkmark$ and exciton diffusion length.

**Decrease** layer thickness: Light absorption  $\downarrow$ Charge collection  $\uparrow$ 

**Increase** layer thickness: Light absorption ↑ Charge collection  $\downarrow$ 

### Al Electrode Al Electrode ITO on Glass / Plastic PCBM MDMO-PP hv Transparent Donor Acceptor contact Photon 1 Solution: Light trapping by plasmonic structures Photon 2 Improving the light absorption by plasmonic effects, enabling to thin down OPVs without sacrificing efficiency, leading to lower material utilization and a cost benefit Useful region ~ 10 nm



## What is a Plasmon?





→A plasmon is a density wave in an electron gas. It is analogous to a sound wave, which is a density wave in a gas consisting of molecules.

→ Plasmons exist mainly in metals, where electrons are weakly bound to the atoms and free to roam. The electrons in a metal can wobble like a piece of jelly, pulled back by the attraction of the positive metal ions that they leave behind. The free electron gas model provides a good approximation (also known as jellium model).

➔ In contrast to the single electron wave function that we encountered already, a plasmon is a collective wave where billions of electrons oscillate in sync.



### **Light Harvesting: Combining Photovoltaics with Plasmonics**





(a) scattering from large diameter (>50 nm) metal NPs into high angles inside the photoactive layer, (b) LSPRs induced by small diameter (5-20 nm) metal particles (c) Excitation of SPPs at the NPs/photoactive layer interfaces ensures the coupling of incident light to photonic modes propagating in the semiconductor layer plane.

E. Stratakis, E. Kymakis, Materials Today (2013), 16, 133-146



## Laser-assisted synthesis of metallic NPs



### Our Approach: Ablation in liquid by ultrashort laser pulses



#### The advantages of laser assisted vs. chemical synthesis

Generation of NPs over a controlled size range with a high degree of reproducibility

Highly soluble in organic solvents

Accurate determination of wt concentration by gravimetric measurements of the target before and after the ablation process

Free of surfactants & passivation layers – no capping layers

 $\rightarrow$  Exciton quenching in the photoactive layer is minimized



Ti-Sapphire laser, 100 femtosecond pulses at 800 nm wavelength



E. Stratakis et al, Optics Express (2009), 17, 12650-12659



## Production on nanoparticles with laser-ablation







## Surfactant free metallic NPs





Ag







Size distribution histogram of NPs calculated from the respective TEM images



## **Surfactant free metallic NPs**





The coloration is assigned to the plasmon oscillation of electrons in the Au, Al nanostructures formed on the target.



### Au NPs are placed into the photoactive P3HT:PCBM layer



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## **Morphological effects**





Incorporation of NPs leads to reduced device degradation rate during prolonged illumination !!!



### Al NPs are placed into the photoactive P3HT:PCBM layer





### Al NPs are placed into the photoactive P3HT:PCBM layer





An efficiency enhancement by 30% is demonstrated, which can be attributed to multiple scattering effects and enhanced structural stability of the photoactive blend.

	$J_{SC}$ (mA/cm <sup>2</sup> )	$V_{OC}(V)$	FF (%)	PCE (%)
Pristine	8.59	0.6	61	3.14
8 % Al NPs	10.60	0.6	60	3.82
9 % Al NPs	11.31	0.6	59	4.00
10 % Al NPs	10.06	0.6	58	3.50





### Au & Al NPs inside the PCDTBT:PC<sub>71</sub>BM photoactive layer





## Au & Al inside the photoactive layer





NPs	J <sub>sc</sub> (mA/cm <sup>2</sup> )	V <sub>oc</sub> (V)	FF (%)	PCE (%)	
No	$11.27 \pm 0.12$	0.86	55.00±0.30	5.33±0.05	
Au	$11.95 \pm 0.20$	0.86	56.00±0.40	5.76±0.09	↑ <i>8%</i>
Al	12.12±0.24	0.86	56.00±0.54	5.84±0.12	↑ <b>9.6%</b>
Au-Al	12.71±0.15	0.86	56.00±0.48	6.12±0.08	↑ <i>14.8%</i>

#### Chem. Commun. (2014), 50, 5285-5287



## Al NPs into PEDOT:PSS buffer layer





Krassas et al., RSC Advances, 5, 71704 – 71708, 2015



## Al NPs into PEDOT:PSS buffer layer



	J <sub>sc</sub> (mA/cm²)	V <sub>oc</sub> (V)	FF (%)	PCE (%)
PEDOT:PSS	10.92 ± 0.12	0.883 ± 0.010	60.0 ± 0.5	5.78 ± 0.12
PEDOT:PSS with 0.6 % Al NPs	11.33 ± 0.10	0.884 ± 0.020	$60.0 \pm 0.4$	6.00 ± 0.12
PEDOT:PSS with 0.8 % Al NPs	11.47 ± 0.15	0.883 ± 0.020	62.0 ± 0.6	6.28 ± 0.16
PEDOT:PSS with 1 % Al NPs	11.13 ± 0.09	0.885 ± 0.010	59.6 ± 0.4	5.87 ± 0.09



#### Krassas et al., RSC Advances, 5, 71704 – 71708, 2015





# Solution processable graphene and 2D-based materials

As electrode, interfacial layer, electron acceptors, and additive materials, compatible with r2r





## **Carbon Nanomaterials**





### Fullerenes – 0D

- Discovered in 1985 (C60)
- C60, C70, C84
- Films n-type semiconducting



### Carbon Nanotubes (CNTs) – 1D

- Discovered in 1991
- Single and multi-walled
- Semiconducting or Metallic



#### <u>Graphene – 2D</u>

- Discovered in 2004
- 2010 Nobel Prize
- Metallic/transparent


### 2010 Nobel Physics Prize !!!





"for groundbreaking experiment s regarding the two-dimensional material graphene"



Molecular structure of graphene



High resolution transmission electron microscope images (TEM) of graphene



# What is so special about Graphene?

- ✓ Atomically thin sheet of carbon
  → flexible, light, abundant material
- ✓ Electrical mobility that is several hundred times that of Si
  - $\rightarrow$  potential for fast electronics
- Transparent conductor
  - ightarrow can combine electrical and optical
    - functionalities High mobility
- Long spin coherence time due to low atomic weight
  - $\rightarrow$  potential for spintronic applications Transpare
- ✓ The strongest material we know
  - ightarrow lightweight composites
- ✓ New material with unique properties
  - → "graphene is probably the only system where ideas from quantum field theory can lead to patentable innovations"







# **Electronic structure of graphene**



Effective mass (related with  $2^{nd}$  derivative of E(k))  $\rightarrow$  Massless Graphene charged particle is massless Dirac fermion.

→ Zero gap semiconductor or Semi-metal





### **Graphene Oxide structure**





Flake sizes: Few microns up to  $100\mu m$ .

- 1. As- Synthesized GO:  $sp^3 = 60\%$
- 2. Solution Processable
- 3. Can be produced in ton quantities

Nature Chemistry 2, 581 (2010)



### **Reduction Approach**



- Graphene oxide sheets are not electrically conductive
- Reduction (de-oxygenation) can be employed to partially restore the graphene network



### **Graphene in OPVs**





# **Graphene in OPVs**



#### Disadvantages of ITO

• Expensive

The cost of indium due to its scarcity and the low throughput deposition process

- Brittleness (not flexible)
  Metal oxides such as ITO (about 150 nm thick to ensure electrical performance) are brittle and therefore of limited use on flexible substrates
- Metal Diffusion
  Indium is known to diffuse into the active layers of OLEDs, which leads to a degradation of performance over time

Looking for other transparent electrodes....

C.Petridis et.al. Nanoscale Horizons, 1, 375 – 382, 2016



## **Pulsed laser in-situ GO reduction**



296 294 292 290 288 286 284 282 2 BINDING ENERGY E\_ / eV





- Chemically exfoliated graphite can be solution processed into graphene-like thin films
- Graphene oxide (GO) can be effectively in-situ reduced by ultrafast UV fs laser irradiation
- Flexible OPVs were successfully demonstrated on spin coated transparent conductive rGO with a comparable performance

Advanced Functional Materials (2013) 23, 2742-2749



### Flexible OPVs with Spin Coated Transparent Graphene Electrodes Reduced by Laser Irradiation









GO thickness (nm)	J <sub>sc</sub> (mA/cm²)	V <sub>oc</sub> (V)	FF	η (%)
4.6	1.77	0.55	0.30	0.29
11.3	4.22	0.57	0.32	0.76
16.4	5.62	0.57	0.34	1.1
20.1	5.56	0.57	0.32	1.01

#### Advanced Functional Materials (2013) 23, 2742-2749



## Laser induced rGOMMs as TCE in OPVs



 A laser-based patterning technique, compatible with flexible substrates, for the production of reduced graphene oxide micromesh (rGOMM) electrodes





# Laser induced Plasmonic rGOMMs as TCE in OP



→ This method involves the irradiation of 2D pristine materials solutions with chloroauric acid (HAuCl<sub>4</sub>) or silver nitrate (AgNO<sub>3</sub>) with short duration UV laser pulses. UV photons absorption activates the surface of the nanosheets and Au/Ag ions are reduced only on the surface of the nanosheet.



GO flakes-pristine solution



GO flakes-Au NPs, laser fluence: 100 mJ/cm<sup>2</sup>



GO flakes-Ag NPs, laser fluence: 100 mJ/cm<sup>2</sup>



➔ Loading graphene flakes with Ag NPs reduces the sheet resistance over an order of magnitude.





# Laser induced Plasmonic rGOMMs as TCE in OP



### Laser induced Plasmonic rGOMMs as TCE in OPVs



# **Graphene based HTL**



Material	Device Structure	PCE	Used as	Ref.
GO	ITO/GO/P3HT:PCBM	3.6%	HTL	ACS Nano (2010), 4, 3169
pr-GO	ITO/pr-GO/P3HT:PCBM/Ca/Al	3.63	HTL	Adv. Mater. (2011), 23, 4923
GO- SWCNTs(1:0.2)	ITO/GO:SWCNTs/P3HT:PCBM/Ca/Al	4.10	HTL	Adv. Energy Mater. (2011), 1, 1052
GO & CO-Cs	ITO/GO/P3HT:PCBM/GO-Cs/Al	3.67	HTL & ETL	Adv. Mater. (2012), 24, 2228
GO & Au NPs	ITO/GO/Au_NPs/P3HT:PCBM/AI	2.9	HTL	Nanoscale (2013), 5, 4144
GO-Cl	ITO/GO-CI/PCDTBT:PCBM/TiOx/AI	6.56	HTL	Nanoscale (2014), 6, 6925



- The plasmonic GO-based devices exhibited a performance enhancement by 30% compared to the devices using the traditional PEDOT:PSS layer.
- They preserved 50% of their initial PCE after **45** *hrs of continuous illumination, contrary to the PEDOT:PSS-based ones that die after 20 hrs*
- Main reason -> Stability improvement



### **Photovoltaic Devices - GO replacing PEDOT**









HTL	V <sub>oc</sub> (V)	J <sub>sc</sub> (mA cm–²)	FF (%)	PCE (%)
ITO only	0.42	6.51	42.5	1.16
PEDOT:PSS	0.60	9.15	51.7	2.86
GO (2.0 nm)	0.43	6.65	36.2	1.04
GO (2.6 nm)	0.54	9.04	43.8	2.09
GO (4.3 nm)	0.60	9.59	50.7	2.90
GO (5.2 nm)	0.60	8.02	44.1	2.12

Nanoscale (2013), 5, 4144-4150



### **GO replacing PEDOT – Stability enhancement**





- The devices with GO as the HTL preserve more than 50% of their initial efficiency after 45 hrs of continuous illumination
- The PEDOT: PSS based devices die after 20 hrs

#### Nanoscale (2013), 5, 4144-4150



### **Combining Plasmonics with Graphene**



GO/NPs

GO

0.5

- • PEDOT-PSS

0.6





- The plasmonic GO-based devices exhibited • а performance enhancement by 30% compared to the devices using the traditional PEDOT:PSS layer.
- They preserved 50% of their initial PCE after 45 hrs of continuous illumination, contrary to the PEDOT:PSSbased ones that die after 20 hrs

#### Nanoscale (2013), 5, 4144-4150

0.3

0.4





### **Doped GO replacing PEDOT**

- R2r compatible photochemical method for the simultaneously • partial reduction and doping of GO films through ultraviolet laser irradiation in the presence of Cl<sub>2</sub> precursor gas
- Photochlorination  $\rightarrow$  grafting of chloride to the edges and the basal • plane of GO
- Tailoring of the GO work-function from 4.9 eV to a maximum value • of 5.23 eV.



a)

I(C-C)/(C-O)

P=50mW

0.3

I(Cl2p)/I(C1s)

### **Doped GO replacing PEDOT**



PCDTBT:PC\_\_BM

0.7

P3HT:PC<sub>61</sub>BM

0.5

0.6

0.8

- OPVs with GO-Cl as the hole transporting layer (HTL) were demonstrated with power conversion efficiency (PCE) of 6.56 %
  - ✓ >14.1 than GO based OPVs
  - ✓ > 15.7 % than PEDOT:PSS based OPVs
- The performance enhancement was attributed to more efficient *hole transportation* due to the energy levels matching between the GO-Cl and the polymer donor.





0.1

0.2

0.3

V (Voltage)

0.4

a)

Current Density (mA/cm²) ଟ ଡ ୨ ୨

14 E

Current Density (mA/cm<sup>2</sup>)

10

0.0

0.1

0.2

- PEDOT:PSS - GO

- GO-CI (5.18 eV)

0.3

0.4

V (Voltage)

0.5

0.6

PEDOT:PSS GO (4.9 eV)

GO-CI (5.04 eV

GO-CI (5.11 eV

— GO-CI (5.17eV) — GO-CI (5.23eV)







#### Chemical functionalization





Chemically functionalized GO-EDNB	J <sub>sc</sub> (mA/cm²)	V <sub>oc</sub> (V)	FF (%)	PCE (%)	
ITO/PEDOT:PSS/P3HT/AI	0.04	0.40	28	0.004	
ITO/PEDOT:PSS/P3HT:GO-EDNB (5%)/AI	2.90	0.67	32	0.62	
ITO/PEDOT:PSS/P3HT:GO-EDNB(10%)/AI	3.32	0.72	40	0.96	ł
ITO/PEDOT:PSS/P3HT:GO-EDNB(15%)/AI	1.78	0.76	28	0.38	
Photochemically functionalized LGO-EDNB					
ITO/PEDOT:PSS/PCDTBT:GO-EDNB(5%)/AI	1.99	1.1	25	0.55	L
ITO/PEDOT:PSS/PCDTBT:GO-EDNB(10%)/AI	3.98	1.09	31	1.34	1
ITO/PEDOT:PSS/PCDTBT:GO-EDNB(20%)/AI	5.29	1.17	39	2.41	
ITO/PEDOT:PSS/PCDTBT:GO-EDNB(30%)/AI	2.91	1.12	29	0.95	
Voltage (V)					

#### Photochemical functionalization









## **2D** Materials-based additives in ternary OPVs







## **2D** Materials-based additives in ternary OPVs





# **2D** Materials-based additives in ternary OPVs



Device structure	Jsc (mA cm <sup>-2</sup> )	Voc (V)	FF (%)	PCE (%)
PCDTBT:PC71BM	10.6±0.13	0.89±0.1	60.2±0.3	5.6±0.1
1.5% (v/v) WS <sub>2</sub>	11.9±0.21	0.87±0.1	59.1±0.2	6.1±0.1
1.0% (v/v) WS <sub>2</sub> -Au	11.1±0.18	0.86±0.1	57.3±0.3	5.4±0.2
1.5% (v/v) WS <sub>2</sub> -Au	12.3±0.22	0.89±0.1	58.4±0.2	6.3±0.1
2.5% (v/v) WS <sub>2</sub> -Au	10.8±0.14	0.86±0.1	57.0±0.4	5.2±0.2



### From lab-cells to mass production



• A proposed means to lower the cost of producing flexible displays in a high-volume manufacturing environment by taking advantage of a unique attribute of flexible substrates relative to the traditional glass substrate



### **Printing solar cells**





*"inks" ---- with electronic functionality!* 







## **Recent research output**

- In situ laser photochemical reduction of graphene oxide for use as the transparent electrode in flexible OPVs
- Chemical functionalization of graphene oxide for use as the electron acceptor or as additives in OPVs
- Combination of graphene oxide with metal nanoparticles for use as the buffer layer in OPVs
- Field emission cathodes based on polymer-graphene and polymer-WS<sub>2</sub> nanotubes electrodes
- Post fabrication laser assisted reduction of GO based FETs





Published Items in Each Year



83 publications 4.100 citations 475,18 Impact points



Center of Advanced Materials & Photonics <u>http://nano.teicrete.gr</u>

> Nanomaterials & Organic Electronics Group





# **Erasmus LLP Organic Electronics & Applications**





**Design and Construction** of the curriculum of a two year MSc joined degree in the modern field of organic semiconductors



### **Erasmus LLP Organic Electronics & Applications**





#### Αριθμ. 3833/Φ120

Έγκριση λειτουργίας του Δι-Ιδρυματικού Μεταπτυχιακού Προγράμματος Σπουδών (Δ.Π.Μ.Σ.), το οποίο συνδιοργανώνεται μεταξύ των Τμημάτων (1) Ηλεκτρολόγων Μηχανικών Τ.Ε. και (2) Ηλεκτρονικών Μηχανικών Τ.Ε. του Τεχνολογικού Εκπαιδευτικού Ιδρύματος Κρήτης, και των Τμημάτων (3) Χημείας και (4) Επιστήμης και Τεχνολογίας Υλικών με του Πανεπιστημίου Κρήτης, με τίτλο "Organic Electronics and Applications" («Οργανικά Ηλεκτρονικά και Εφαρμογές»).

http://orea.chania.teicrete.gr

and Laser

nstitute of Electronic Structure

(3)





### **TEI** of CRETE

The objectives of this Erasmus Plus KA2 Strategic

- Introduce to students & academics the modern

Inspire the first year postgraduate student's

Trigger the introduction of modern educational

curricula in the field of Electronic Engineering

**Bey**ond **Si**licon **Er**a are the followings:

students and academics in Europe

ELBYSIER **Electronics Beyond Silicon Era** 



### **Partners**

- **University of Crete GR**
- **New University of Lisbon PT** ٠
- **University of Bucharest RO** •
- **University of Warsaw PL**
- École des Mines St Etienne FR •
- **TEI of Crete GR**

### **Intellectual Outputs**

**Educational Programs for final year undergraduate &** first year postgraduate students in

- **Organic Electronics**
- **Transparent & Flexible Electronics**
- **Bioelectronics** •
- **Nanoelectronics**
- **Spintronics** •

Nanomaterials & **Organic Electronics Group** 





# **Forthcoming Event**

trends of electronics

research

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**Intensive Course in Transparent & Flexible Electronics** Chania, Crete, GR, 9-14 October 2016

Info: https://elbysier.chania.teicrete.gr/







# **Special achievements**

- Graphene Flagship funding and deputy leader in the work package of energy in the **EU's biggest research initiative ever.**
- Other national and international funds



**GRAPHENE FLAGSHIP** 

 Design and Construction of the curriculum of a two year MSc joined degree in the modern field of organic semiconductors





### **Nanomaterials & Organic Electronics Group**



#### **Prof. Emmanuel Kymakis**

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**Collaborators** Dr. Emmanuel Stratakis, Ultrafast Laser Micro- and Nano- processing group of IESL of FORTH

Dr. Vittorio Pellegrini and Dr. Francesco Bonaccorso, Istituto Italiano di tecnologia, Graphene Labs

**Center of Advanced Materials & Photonics** 

http://nano.teicrete.gr/







# **Special Thanks**







# Thank you for you attention!!!

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22 - 24 September, 2016 Heraklion, Crete, Greece



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